

Report on Comparison Measurement

Phosphate buffer 2010 / 2011



1 Objectives

The comparison measurement was carried out in preparation of application for accreditation / CMC entries for the pH calibration laboratory of SNM INDECOPI.

2 Participants

Pilot laboratory DKD-K-06901 ZMK -ANALYTIK- GmbH Ortsteil Wolfen P-D ChemiePark Bitterfeld-Wolfen Areal A, Filmstr. 7 06766 Bitterfeld-Wolfen *Contact person: Diana Jehnert (Deputy Head of DKD-K-06901) e-mail: info@zmk-wolfen.de*

Participant Servicio Nacional de Metrología – INDECOPI Especialista 2 Lima, Peru *Contact person: Galia Ticona e-mail: gticona@indecopi.gob.pe*

3 Time schedule

Start of comparison measurement:	October 2010		
Submission of results by participant:	March 2011		
Evaluation of results:	April 2011		



4 Calibration object

Measuring object:	pH buffer solution / Phosphate
Nominal value:	рН 7
Prepared by:	ZMK -ANALYTIK- GmbH
Lot No.:	F1589786936 // A0135473017
Bottle No. (Participant)	13 and 14
Date of bottling:	15 th October 2010
Date of calibration (by ZMK):	18 th October 2010
Measuring temperature:	25°C

Homogeneity of the batch was determined before the comparison measurement was started. SNM INDECOPI received 2 bottles of 250 ml (bottle no. 13 and 14).

5 Applied calibration methods

5.1 DKD-K-06901 / ZMK -ANALYTIK- GmbH

For the determination of the reference value the *Differential potentiometry* was used by the pilot laboratory. The measuring cell is constructed according to Baucke [1]. The electrode system consists of two platinum electrodes that are surrounded by hydrogen.

The measuring cell is placed in a thermostatic bath with known stability and homogeneity.

A certified reference buffer solution was used as standard (Lot. No. 81680/73050, Batch No. 32, U(k=2)=0.003).

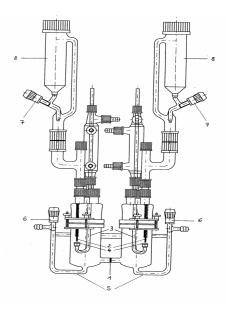


Fig. 1 Baucke cell design

[1] F.G.K. Baucke, J. Electroanal. Chem. 368 (1994), 67-75



5.2 SNM INDECOPI

The participant took part in the comparison measurement with two calibration methods.

Differential potentiometry

The measuring cell is constructed according to Baucke [1]. The electrode system consists of two platinum electrodes that are surrounded by hydrogen. The measuring cell is placed in a thermostatic bath with known stability and homogeneity. The potential difference was measured with a precision multimeter (Agilent 34420A).

A certified reference material with an uncertainty of U(k=2)=0.003 was used as standard: Phosphate Lot. No. 73050/81680

Fig. 2 Baucke cell of SNM INDECOPI

certificate No.: 10-0241/DKD-K-06901/09-02

Multipoint calibration by glass electrode system

The calibration was carried out using a glass electrode measuring system.

Glass electrode:	Туре:	Inlab Science	
	Manufacturer:	Mettler Toledo	
pH Meter:	Туре:	PP50	
	Manufacturer:	Sartorius	
	Resolution:	0.1 mV	



The following five certified reference materials were used as standards:

Tetraoxalate	Lot. No. 41908152	certificate No.: 10-0239/DKD-K-06901/09-02
Phthalate	Lot. No. A915674743	certificate No.: 10-0240/DKD-K-06901/09-02
Phosphate	Lot. No. 73050/81680	certificate No.: 10-0241/DKD-K-06901/09-02
Borate	Lot. No. 71840	certificate No.: 10-0242/DKD-K-06901/09-02
Carbonate	Lot. No. DMR-324-Ia / DMR-324-IIa	certificate No.: CNM-MR-630-0459/2009

The EMF of each reference solution and the sample was measured. Using the calibration results a regression line and its parameters were determined. These parameters were used to calculate the pH value of the unknown sample. The calibration was carried out in a water bath with known stability and homogeneity.

6 Measuring results

The calibration results were reported in evaluation excel-sheets stating all raw data, calculation results and measuring uncertainties. The results of the comparison measurement are summarized in table 1.

In order to compare the results the value of E_n was calculated (see equation 1).

$$E_{n} = \frac{\left| x_{lab} - x_{ref} \right|}{\sqrt{U(x_{lab})^{2} + U(x_{ref})^{2}}}$$
(1)

This value represents the deviation between the measuring results x_{lab} and x_{ref} of the participating laboratory and the pilot laboratory under consideration of the measuring uncertainty. For acceptable measurements the value of E_n must be less than 1.



	ZMK (DKD-K-06901)		SNM INDECOPI				
Measuring temperature	Calibration method	Reference pH value <i>x_{ref}</i>	exp. uncertainty <i>U_{ref}</i>	Calibration method	pH value <i>x_{lab}</i>	exp. uncertainty <i>U_{lab}</i>	En
25 °C	Differential potentiometry	6.866	0.003	Differential potentiometry	6.865	0.003	0.24
25°C	Differential potentiometry	6,866	0.003	Multipoint calibration	6,87	0.02	0.20

Table 1: comparison measurement results

7 Summary

The determined E_n value for the comparison measurement on pH is less than 0.3. The comparison measurement was passed successfully by the participant SNM INDECOPI.

Bitterfeld-Wolfen, 29th April 2011

Diana Jehnert Deputy Head of DKD-K-06901 Dr. Barbara Werner Head of DKD-K-06901